

List of practicals in Nuclear and Particle Physics (DSE-II) in Scilab.

- 1) Nuclear Radius.** Plot nuclear radii as a function of its atomic mass number. That is plot R vs A . Since nucleon density is constant over the nucleus (nucleon are uniformly distributed over the nucleus) $R = R_0 A^{1/3}$. Take $R_0 = 1.2$ fm.
- 2) Nuclear Charge Distribution.** Plot the following nuclear charge distribution: $\rho = \rho_0 e^{-\ln(2)\left(\frac{r}{R}\right)^2}$. That is plot ρ/ρ_0 with r , r given in units of R , the mean nuclear radius; $R_0 A^{1/3}$.
- 3) Nuclear Form Factor.** Evaluate the nuclear form factor $F(q)$ based on the above charge distribution, $\rho = \rho_0 e^{-\ln(2)\left(\frac{r}{R}\right)^2}$. Plot the form factor $F(q)$ as a function of q ; where q is the change in momentum due to scattering, $F(q) = \frac{4\pi}{q} \int \sin(qr') \rho(r') r' dr'$.
- 4) Mass Parabola.** The minimum atomic number in the mass parabola is given by $z_{\min} \approx \frac{A}{2} \left(1 + \frac{1}{4} A^{\frac{2}{3}} \frac{a_c}{a_{\text{sym}}} \right)^{-1}$. Plot z_{\min} vs A . Take $a_c = 0.72$ MeV and $a_{\text{sym}} = 23$ MeV. Do you find $z_{\min} \sim A/2$ for small A and $< A/2$ for large A ?
- 5) Semi-Empirical Mass Formula.** Plot B/A vs A for any Z . (Say $Z = 56$) $B = a_v A - a_s A^{\frac{2}{3}} - a_c Z(Z - 1) A^{-\frac{1}{3}} - \frac{a_{\text{sym}}(A - 2Z)^2}{A} + \delta$. Take $a_v = 15.5$ MeV, $a_s = 16.8$ MeV, $a_c = 0.72$ MeV, $a_{\text{sym}} = 23$ MeV, $a_p = 34$ MeV and $\delta = 0$ for odd A , $\delta = a_p A^{-3/4}$ for even N , even Z and $\delta = -a_p A^{-3/4}$ for odd N , odd Z . Plot separately: a. volume terms only, b. volume + surface terms, c. volume + surface + coulomb terms, and finally d. volume + surface + coulomb + symmetry terms.
- 6) Nuclear Potential Energy.** Plot V assuming a point nucleus and assuming uniform spherical charge distribution. $V_{pc} = -\frac{Ze^2}{4\pi\epsilon_0 r}$. $V_{sph} =$

$-\frac{Ze^2}{4\pi\epsilon_0 R} \left\{ \frac{3}{2} - \frac{1}{2} \left(\frac{r}{R} \right)^2 \right\}$. For convenience treat $\frac{e^2}{4\pi\epsilon_0} = 1$ and chose scale of r in suitable range of R.

7) Range of Force. Plot Range of a force R vs mass of its carrier particle

$$m_x c^2. R = c\Delta t = \frac{hc}{2\pi m_x c^2} = \frac{197.3 \text{ MeV.fm}}{m_x c^2}.$$

8) Mean radius squared. Plot mean of radius squared vs A, (i.e. expectation value of radius squared vs A) for the nucleus, assuming nucleus

$$\text{as a uniform charged sphere. } \langle r^2 \rangle = \frac{3}{5} R^2 = \frac{3}{5} R_0^2 A^{\frac{2}{3}}.$$

9) Gaussian Probability Distribution. Plot the Gaussian probability distribution for a standard deviation $\sigma = 0.01$ and mean $\mu = 0.1$. $P =$

$$\frac{1}{\sqrt{2\pi\sigma^2}} e^{-\frac{(x-\mu)^2}{2\sigma^2}}.$$

10) Kinetic energy of alpha particle. Plot Kinetic energy of alpha particle T_α vs mass number A. Assume Q value is 5 MeV. $T_\alpha = Q \left(1 - \frac{4}{A} \right)$.

11) Neutrino Mass. Plot variation in Q if $m_\nu c^2$ varies between 0.01 to 0.08 MeV. $Q = 0.782 \text{ MeV} - m_\nu c^2$. For the decay $n \rightarrow p + e^- + \text{anti-}\nu$.

12) Power radiated by accelerated charge. Plot power radiated (P) by an electric dipole of unit strength, depicted by the following relation, vs

$$\text{frequency, } \omega. P = \frac{1}{12\pi\epsilon_0} \frac{\omega^4}{c^3} d^2.$$

13) Scattering Cross Section. Plot Rutherford differential scattering cross section vs sine of scattering angle, θ . $\sigma_{diff} =$

$$\frac{d\sigma}{d\Omega} = \left(\frac{zZe^2}{4\pi\epsilon_0} \right)^2 \left(\frac{1}{4T_\alpha} \right)^2 \frac{1}{\sin^4 \frac{\theta}{2}}. \text{ Use KE of alpha particle } = T_\alpha = 10 \text{ MeV}.$$

Use Gold nucleus for Z.

14) Fusion Reaction. Plot plasma Debye length L_D vs particle concentration n. Take n in units of $10^{27} / \text{m}^3$. Mean KE per particle = 10 keV.

$$T \text{ at the order of } 10^8 \text{ K. } L_D = \frac{4\pi\epsilon_0}{e^2} \sqrt{\frac{kT}{4n}}.$$

15) Synchrotron Condition. Plot ν (frequency) vs B (magnetic field)

for protons at radius $r = 0.25$ m. $\nu = \frac{eBc^2}{2\pi} \sqrt{e^2 r^2 B^2 c^2 + m^2 c^4}$.